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**LETTERS
TO THE EDITOR**

Reaction of 2-Bromo-1,3,2-Dioxaphospholane with Salicylalethanediamine

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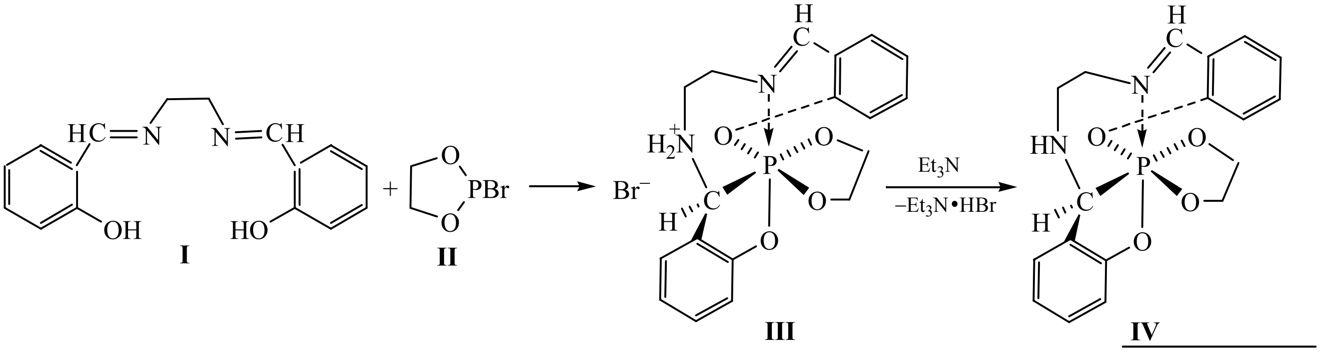
We have previously shown that the reaction of salicylalethanediamine with some cyclic chlorphosphites results in hexacoordinated phosphorus atom derivatives involving the intramolecular transannular N→P bond (according to the X-ray diffraction method) [1]. Aiming to study the synthetic opportunities of this reaction, we carried out for the first time the reaction of ethylenebromophosphite **II** with bisazometine **I**. According to the ^{31}P NMR data, the reaction occurs with a high stereoselectivity to form diastereomerically pure product **III** in almost quantitative yield. The phosphorus chemical shift of this product is δ_P –118.05 ppm, which is characteristic for the P(VI) derivatives.



At treating with triethylamine, product **III** undergoes dehydrobromination to transform into a neutral phosphate **IV** (δ_P –101.52 ppm). The structure of the compounds obtained was confirmed by the ^{1}H and ^{31}P NMR, IR spectroscopy, mass spectrometry and elemental analysis data.

1,1-Ethylenedioxy-3,4,11,12-dibenzo-6-aza-9-ammonium-2,13-dioxa-1-phospha[8.3.0^{1,10}]tridecatri-3,5,11-ene bromide (III). To a solution of 0.27 g diimine **I** in 5 ml of methylene chloride was added 0.17 g of bromide **II**. A self-heating of the reaction mixture was observed. After 4 h, the precipitate formed was separated and washed with diethyl ether. Yield 0.42 g (96%), mp 282°C. IR spectrum, ν, cm^{-1}: 1625 (C=N), 2415–2678 (NH_2^+). ^{1}H NMR spectrum (CD_3Cl), δ, ppm: 3.54–4.07 m (8H, NCH_2, OCH_2), 4.46 d (1H, PCH, $^2J_{HP}$ 7.85), 6.86–7.71 m (8H, CH_{Ar}), 8.72 d (1H, CHN=, $^3J_{PH}$ 7.62). ^{31}P NMR spectrum (DMSO-d_6): δ_P –118.05 ppm. Mass spectrum (MALDI-TOF), *m*/*z*: 358 [M^+ – HBr]. Found, %: N 6.01; P 7.06. $C_{18}H_{20}N_2O_4PBr$. Calculated, %: N 6.38; P 7.05.

1,1-Ethylenedioxy-3,4,11,12-dibenzo-6,9-diaza-2,13-dioxa-1-phospha[8.3.0^{1,10}]tridecatri-3,5,11-ene (IV). A mixture of 0.44 g of compound **III** and 0.1 g of triethylamine in 10 ml of methylene chloride was refluxed for 2 h. The solvent was distilled off, the reaction mixture was dissolved in benzene, and triethylamine hydrochloride was filtered off. The resulting solution was evaporated. Yield 0.27 g (74%), mp 177°C. IR, ν, cm^{-1}: 1635 (C=N), 3345 (NH). Mass spectrum (MALDI–TOF), *m*/*z*: 358.3. ^{31}P NMR spectrum (CD_3Cl): δ_P –105.48 ppm. Found, %: N 7.49; P 8.31. $C_{18}H_{19}N_2O_4P$. Calculated, %: N 7.81; P 8.53.

The ^1H NMR spectra were recorded on an Avance 600 spectrometer (600.13 MHz, $CDCl_3$). The ^{31}P NMR spectra were recorded on a Bruker Fourier MSL-400 spectrometer (100.62 MHz). The IR spectra were recorded on a Vector 22 Fourier spectrometer (Bruker) in the range 400–3600 cm^{-1} from KBr pellets. The mass spectra (MALDI–TOF) were obtained on an Ultraflex III TOF/TOF Bruker instrument (matrix *p*-nitroaniline).

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